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tration of the acid? A 50.0-mL sample of 0.0100M is neutralized by 45.6 ml. of HBr. What is the molarity Of the acid? 9. What is its  $[H^+]$ ? 10. What is its pH? 11. What is its pOH? A solution has a  $[1-14]$  of  $5.6 \times 10^{-6}$ M. Chemistry: Matter and Change Chapter 19

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